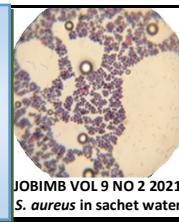


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Nitrogen Fixing Bacteria and Their Application for Heavy Metal Removal: A Mini Review

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ABSTRACT

Nitrogen is a critical component of biological systems and typically serves as a constraint on production in both aquatic and terrestrial environments, although its shortage has been compensated for through the process of biological nitrogen fixation. Nitrogen fixation is a critical microbial activity that utilises nitrogenase enzymes to convert dinitrogen (N₂) gas to ammonia (NH₃). It is carried out by a diverse spectrum of bacteria known as nitrogen fixing bacteria. These include free-living bacteria such as *Azotobacter*, *Bacillus*, *Beijerinckia*, and *Clostridium*, associative bacteria such as *Azospirillum*, *Enterobacter*, and *Pseudomonas*, and bacteria that form symbiotic associations with legumes such as *Rhizobium* and actinorrhizal plants such as *Frankia*. These bacteria contribute significantly to plant growth by producing phytohormones (such as auxins, cytokinins, gibberelins, and indole acetic acid), reducing the incidence of plant diseases through the production of siderophores and cell wall degrading enzymes, and increasing phosphorus nutrition via phosphate solubilization. Additionally, they remove heavy metal ions from solutions through a process called biosorption, which is a feasible, natural, environmentally benign, and economically viable technique of remediating heavy metal-contaminated environments.

INTRODUCTION

Nitrogen is the most important nutrient for plant development and production in agribusiness[1]. To sustain life, it is one of the most crucial elements; a primary source for protein, nucleic acid, and other organic nitrogenous substances synthesis. Chemically, nitrogen fixation occurs when airborne molecular nitrogen (N₂) is transformed into ammonia (NH₃) or other nitrogen compounds, most often in soil or aquatic systems [2], although it may also be used for industrial purposes. Most microbes can't utilise atmospheric nitrogen since it is molecular dinitrogen, an inert and nonreactive chemical. Nitrate is converted to ammonia (NH₃) by the nitrogenase protein complex in a microbiological process known as diazotrophy or nitrogen fixation (Nif)[3]. In contrast, the process of nitrogen mineralization in the soil is relatively sluggish, accounting for just one to three percent of the total soil nitrogen[4].

There appears to be a large number of heterotrophic bacteria living in soil, and these bacteria appear to be responsible for major nitrogen reductions [5]. One of the most significant microorganisms in agricultural soil is the nitrogen-fixing bacteria (also known as nitrobacteria) [6]. Certain species of *Enterobacter*, *Azospirillum*, and *Pseudomonas* have been shown to be mutualistic or symbiotic with diverse types of plants, including *Frankia*, *Rhizobium*, and certain *Azospirillum* and *Enterobacter* strains [7]. Soil microorganisms known as diazotrophs, including bacteria such as *Azotobacter* and archaea, fix nitrogen organically. Plant groupings, particularly legumes, have symbiotic partnerships with nitrogen-fixing bacteria. When nitrogen is fixed by rice roots, the link between diazotrophs and plants is sometimes referred to as associated relationship or symbiotic. Some termites and fungi are able to fix nitrogen via a process known as nitrogen fixation. [8] Lightning-induced NO_x emissions cause it to spontaneously arise in the atmosphere. Because of their capacity to convert nitrogen from the

atmosphere into a form that plants can utilise, these bacteria are among the most well-known [9]. As a result, it is imperative that the world's resources are tapped as quickly as possible, and the process of biological nitrogen fixation is a key component. A method for converting atmospheric nitrogen into fixed nitrogen is biological nitrogen fixation [10].

Bacteria that fix nitrogen facilitate this transition [11]. Many nitrogen-fixing bacteria have been isolated from various environments, including bacteria from the genera *Azospirillum*, *Bacillus*, *Azotobacter*, *Clostridium*, and Onyeze et al. [12], *Klebsiella* [13], and *Rhizobium* [14], all of which were isolated from soil [12,13,15] and from compost [16]. Additional to this, these bacteria are capable of promoting plant growth by synthesising plant growth promoting hormones such as cytokinins, auxins, and geberellins, inhibiting phytopathogenic organisms through production of fungal cell wall degrading enzyme, decreasing disease incidence through the secretion of antibiotic-like substances, and increasing phosphorus nutrition by phosphate solubilization [4]. This study provides an overview of the major group of bacteria known as nitrogen fixing bacteria in light of their contribution.

Biological Nitrogen Fixation

The principal means by which vegetation in native habitat get nitrogen necessary for development is through a process known as biological nitrogen fixation (BNF) [17]. Nearly one-third of the biologically fixed nitrogen (BNF) in terrestrial and aquatic ecosystems is produced by a very diversified group of Prokaryotes (Bacteria and Archaea) known as diazotrophs [18, 11]. In addition to free-living bacteria from the *Azotobacteria*, *Burkholderia*, *Bacillus*, *Azospirillum* and *Clostridium* genera, there are symbiotic bacteria linked with *Rhizobium*, actinorhizal plants like *Frankia*, and cyanobacteria connected with the cycad genus *Cyanobacteria* [19]. It is only methanogens that can fix nitrogen in Archaea [20]. *nifH*, a gene encoding a subunit of the iron protein of nitrogenase, is highly conserved throughout nitrogen-fixing groups and is an appropriate molecular identifier for these bacteria. Nitrogen-fixing bacteria express the nitrogenase enzyme complex [21].

Researchers researching the phylogeny, variety, and abundance of nitrogen-fixing bacteria use it as a marker gene of choice [22]. The only known natural mechanism for reducing N_2 to NH_3 is the enzyme complex found in bacteria, called bacterial nitrogenase [23]. The fixation of one molecule of nitrogen into two molecules of ammonia by BNF is one of the most expensive metabolic activities, requiring 16 molecules of ATP [23]. BNF is a cost-effective and environmentally friendly method for achieving long-term agricultural output sustainability [24].

Biological Associations of Nitrogen Fixing Bacteria

Symbiotic Nitrogen Fixing Bacteria

To fix biological nitrogen, heterocystic cyanobacteria and cycads [19], *Frankia* and the rhizome nodules of woody nonlegumes [25], as well as rhizobial bacteria (such as *Rhizobium*) and legumes (alfalfa, soybean) form symbiotic associations. This is how biological nitrogen fixation works [16]. When natural nitrogen is scarce, the symbiotic relationship between rhizobial bacteria and legumes has been shown to satisfy nitrogen demands for proliferation [26]. It is among the microorganism relationships that has been investigated the most thoroughly and evaluated the most exhaustively [27]. Legumes and Rhizobia work together to create Nod factors, which are then signalled by flavonoids released by the roots of legumes. These Nod factors aid plants in identifying microorganisms. Nodules are formed when legumes engage with rhizobia, resulting in bacterial

infection. Rhizobia can fix nitrogen in the nodule because it has a low oxygen content [28, 29]. The legumes' carbon is exchanged for nitrogen fixed by the rhizobia inside the nodules [30].

Non-Symbiotic Nitrogen Fixing Bacteria

Bacteria that fix nitrogen without forming a symbiotic relationship with plants are known as non-symbiotic nitrogen fixers and can be found in soil or water, either in connection with plants or as free-living organisms.

Associative Nitrogen Fixing Bacteria

Bacteria belonging to the genera *Azospirillum*, *Paenibacillus* and *Herbaspirillum* [31], *Klebsiella* [13], *Pseudomonas* [32], and *Enterobacter* execute associative nitrogen fixation [33]. Free-living and symbiotic nitrogen fixers are separated by the associative nitrogen fixers in terms of the amount of nitrogen fixed [34]. On the roots of corn, wheat, and sugarcane, they have been most commonly found, according to the literature [35]. As their exudates include carbon, they aid the associated bacteria in obtaining a supply of carbon for nitrogen fixation [36].

Free-living Nitrogen Fixing Bacteria

Symbiotic bacteria play an important role in nitrogen fixation; nevertheless, free-living nitrogen fixers provide a significant quantity of nitrogen to ecosystems [37]. Proteobacteria, Firmicutes, Archaea and Cyanobacteria are only few of the heterotrophs that may convert nitrogen to nitrous oxide in the absence of a food source [38]. The oxidation of organic molecules provides the energy for this process. In compared to symbiotic nitrogen fixation, free-living diazotrophs are thought to provide a very small quantity of total fixed nitrogen. The free-living nitrogen fixers fix around one-tenth of the total atmospheric nitrogen fixed by the symbiotic relationship [24, 37].

Symbiotic nitrogen fixers fix more nitrogen than free-living and associated nitrogen fixers, on the other hand [29]. Symbiotic nitrogen fixation utilises the host plant as a carbon source of power and a shield against oxidation process damages of the nitrogenase enzyme [29]. In the absence of oxygen, anaerobes and facultative anaerobes convert nitrogen from the atmosphere into ammonia. It is possible for *Azospirillum* bacteria to fix nitrogen in environments with low oxygen levels. Nitrogen fixing in aerobic bacteria like *Azotobacter* happens by reducing the intracellular oxygen content. The heterocyst of cyanobacteria is where nitrogen fixing takes place [39]. With a rise in oxygen in the atmosphere, nitrogen fixation reduces [37].

Plant Growth Promotion by Nitrogen Fixing Bacteria

Plant growth stimulants have been linked to nitrogen-fixing microorganisms. Using both direct and indirect mechanisms, they encourage plant development [40]. Nitrogen fixing bacteria boost plant development by releasing phytohormones such as indole acetic acid (IAA), cytokinin's [41], gibberellins [42], and auxins [42] in addition to the nitrogen fixation described above [43, 44]. Direct mechanisms include release of a molecule lessening phytopathogenic organism effects and producing siderophore (low molecular weight iron chelating compounds), cell wall disintegrating enzymes such as chitinase, and antibiotic-like substances such as sulfonamide [45].

Solubilization of insoluble phosphorus by *Rhizobium* bacteria and species of *Azotobacter* have been shown to boost plant development in both legumes and non-legumes, respectively [46]. These bacteria are more beneficial than free-living or associative nitrogen fixing bacteria because they are well protected within the nodule tissue and encounter little or no competition from indigenous soil microorganisms. Production of

indoleacetic acid by *Rhizobium*, *Azotobacter* and *Bacillus*, *Paenibacillus* and *Pseudomonas* has been reported [47, 48] and plays a central role in plant growth and development and acts as a signalling molecule which is involved in plant signal processing and motility or signalling [49] *Alcaligenes* sp. [49] has also been reported to produce indoleacetic acid [50]. *Pseudomonas* [51,49] and *Azotobacter* [52,53] have all been shown to produce siderophore (a low molecular weight iron chelating compound). *Azospirillum* [54] and *Rhizobium* [15, 55]. It has been shown that phytopathogens are suffocated by siderophores because they are unable to acquire iron from the soil, and this has been linked to siderophores [40].

Gibberellin is a key phytohormone in the growth and functioning of plants. *Bacillus* species such as *Bacillus cereus* MJ-1, *Bacillus macroides* CJ-29, and *Bacillus pumilus* have been shown to produce gibberellin [42]. Cytokinins [41], a phytohormone that regulates cytokinesis, growth, and development in plants, have been shown to be produced by nitrogen-fixing bacteria [56]. Nitrogen-fixing *Bacillus megaterium* [41] produced cytokinin (UMCV1). *Bacillus* sp. has also been shown to produce antibiotics and to degrade cell walls [57]. Antibiotics are made up of a variety of low-molecular-weight secondary metabolites that are harmful to other microorganisms, such as plant pathogens, as well as to themselves [58]. Nitrogen-fixing *Bacillus* has produced several antibiotics, including amphisin, 2,4-diacetylphloroglucinol (DAPG), hydrogen cyanide, phenazine, and pyrolnitrin [57, 58]. Growth-promoting properties of the cell wall-degrading enzymes generated by several nitrogen-fixing bacteria, most often *Bacillus*. It is possible to destroy the cell wall of many phytopathogenic mushrooms with enzymes such as chitinase, cellulase, beta 1,3 glucanase, protease and lipases, reducing the frequency of plant illnesses [59].

Biosorption of Heavy Metals by Nitrogen Fixing Bacteria

The importance of heavy metal pollution remediation technology cannot be overstated in a community that is able to meet both its economic and environmental needs. It's costly, requires a lot of chemicals, and results in harmful chemical sludge when using conventional techniques to remove metals from aqueous solutions [60]. Natural and ecologically friendly adsorbents, such as biomass from microbes and plants, are essential for metal adsorption, a process known as "biosorption." Toxic metal levels can be reduced to ecologically reasonable standards through biological therapy based on microbial and plant biomass [61]. Toxic metal ions have been removed from various matrices using microbial biomass [61-63].

Microbial biomass materials have been demonstrated to successfully adsorb heavy metals from even extremely dilute aqueous solutions, and this has implications for the biosorption of heavy metals from wastewater by biological materials [64]. Complexation, coordination, physical adsorption, chelation, ion exchange, inorganic precipitation, or a mix of these activities are some of the ways metal biosorption mechanisms operate [65]. For microorganisms to remove metal through biosorption mechanisms, several variables must come into play, including metal ion composition and ionic strength as well as cell wall composition, physiology, and physicochemical conditions like pH and temperature [62,63].

Various heavy metals can be removed or absorbed from solutions by different kinds of nitrogen-fixing bacteria and their products. Despite their importance in plant host specificity [66, 67] and growth promotion, rhizobial exopolysaccharides have only lately been studied for their metal sorption capability. At

concentrations of 15.5, 20 and 25mg, *Azotobacter's* extracellular polymer was able to bioadsorb Cu, Zn, and Fe. After the Chernobyl nuclear disaster, researchers Douka and Xenoulis [68] found that nodulated pasture legumes significantly reduced radioactive metal concentrations. *Rhizobium trifolii* was reported to lower UO_2 2 contents by 60% in a 0.4 mM solution with a maximal sorption capacity of 0.25 mmol UO_2 2+/g, according to Cotoras et al.[68]. Using *Azotobacter chroococcum* XU1 exopolysaccharide, Rasulov et al. [70] explored the biosorption of metal ions such as lead (Pb) and mercury (Hg) from solution and observed substantial removal of these ions from solution. Chromium (Cr) sorption by *Azotobacter* s8, *Bacillus subtilis*, and *Pseudomonas aeruginosa* live cells was studied by Kurniawan et al. [71]. Adsorption capabilities of 135.3 and 167.5 mg/g for cadmium and cobalt, respectively, were achieved by *Rhizobium leguminosorum* in research by Abd-Alla et al.[72]. N_2 fixation bacteria were found to create polysaccharides and other biopolymers that have metal-binding capabilities [73]. Polysaccharides such as peptidoglycans, water-soluble and amphipathic exopolysaccharides (EPS), teichoic and teichuronic acids as well as capsular polysaccharides and polyglutamic acid are among the most common (LPS). If the cations are bound to the bacteria via electrostatic interactions, they are usually interacting with negatively charged functional groups such as uronic acids (EPS from *Bradyrhizobium japonicum* or alginate or teichuronic acid or emulsan or LPS from various sources), membrane phosphoryl groups or carboxylic amino acid groups. Cation binding by positively charged polymers [74] or coordination with hydroxyl groups may also play a role in electrostatic interactions[75, 76]. Eukaryotic polymers, such as chitin and chitosan, have been shown to bind cations, perhaps through chelation and coordination with hydroxyl groups [74]. LPS attaching metal ions to cell walls may not be directly linked to O-antigen side chains, but rather, the B-band LPS may alter cell surface characteristics, allowing metals to precipitate in certain areas of cell surface, as hypothesised by Langley and Beveridge[77].

For years, scientists are looking at the possibility of using microorganisms like bacteria that can resist, detoxify, and absorb metals to clean up metal-polluted settings [78]. Biosorbents and biosorption systems that are efficient, natural, eco-friendly, and financially feasible may be one answer to bioremediation of settings contaminated by heavy metals [79].

CONCLUSION

Nitrogen-fixing bacteria convert atmospheric nitrogen into plant-available form and are essential to soil health and plant growth. Plant growth promotion is promoted by the synthesis of various hormones and enzymes as well as the inhibition of phytopathogens by these microbes. A process known as biosorption provides an efficient, natural, eco-friendly, and economically viable method of removing heavy metal ions from solutions that are polluted with them. Reviewers found that these particular microbes have a variety of responsibilities in maintaining a healthy ecosystem.

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