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Molybdenum Blue Production from *Serratia* sp. strain DRY5: Secondary Modeling

Ibrahim Alhaji Sabo^{1,3}, Salihu Yahuza^{2,3}, and Mohd Yunus Shukor³*

¹Department of Microbiology, Faculty of Pure and Applied Sciences, Federal University Wukari, P.M.B. 1020 Wukari,

Taraba State Nigeria.

²Department of Microbiology and Biotechnology, Faculty of Science, Federal University Dutse, P.M.B., 7156, Dutse,

Jigawa State, Nigeria.

³Department of Biochemistry, Faculty of Biotechnology and Biomolecular Sciences, Universiti Putra Malaysia, 43400 UPM Serdang, Selangor, Malaysia.

*Corresponding author: Prof. Dr. Mohd. Yunus Abd. Shukor Department of Biochemistry, Faculty of Biotechnology and Biomolecular Sciences, Universiti Putra Malaysia, 43400 UPM Serdang, Selangor, Malaysia.

Email: <u>yunus.upm@gmail.com</u>

ABSTRACT

HISTORY

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INTRODUCTION

Heavy metals are necessary for the biological processes of plants and animals, but at high quantities, they can disrupt metabolic reactions in organisms' systems. Toxic heavy metals, such as molybdenum, which is beneficial to plants, can impair plant development by lowering photosynthetic activities, plant mineral nutrition, and necessary enzyme function [1]. Metals are used in a wide range of industrial processes. As a result, many Malaysian environmental systems have been contaminated with heavy metals [2]. Due to indiscriminate dumping and unlawful discharge, heavy metal pollution has become a public health concern in Malaysia [3]. Toxic heavy metals, such as molybdenum, can build up in the body after being absorbed in contaminated food and pose a health danger to living organisms [1]. Molybdenum is a heavy metal with a wide range of industrial applications. It's a dangerous contaminant, with thousands of parts per million levels found in soils and aquatic bodies [2]. Molybdenum has also been shown to suppress spermatogenesis in drosophila and fish [4]. Molybdenum is a common metal that is frequently contaminated by mining operations and human sources. Molybdenum is becoming a global contaminant at an alarming rate. Its contamination levels have been measured in water and soils across the world, including Terengganu in

Monod, Han and Levenspiel were used to model molybdenum blue production from *Serratia* sp. strain DRY5. Based on statistical analyses such as root-mean-square error (RMSE), adjusted coefficient of determination $(adjR^2)$, bias factor (BF), and accuracy factor (AF), the Monod model was chosen as the best. The calculated values for the Monod constants q_{max} (the maximum specific substrate degradation rate (h^{-1}) , and K_s (concentration of substrate at the half maximal degradation rate (mg/L)) were found to be 3.86 (95% confidence interval of 2.29 to 5.43), and 43.41 (95% confidence interval of 12.36 to 74.46) respectively. The novel constants discovered during the modelling exercise could be used in further secondary modelling.

In this work, kinetic growth models such as Luong, Yano, Teissier-Edward, Aiba, Haldane,

Malaysia, Tokyo Bay, Tyrol in Austria, and the Black Sea, where molybdenum levels have reached dangerously high levels [3,5,6]. It is non-toxic to humans; however, it is lethal to ruminants at concentrations of many parts per million [5]. Molybdenum is used in the production of steel, corrosionresistant steel components, engine anti-freeze additives, and molybdenum disulfide in lubricants, among other things. The usage of molybdenum in the industry has resulted in several examples of soil and water contamination [7]. In general, the reduction mechanism in molybdenum-reducing bacteria has been shown to involve an enzymatic reaction rather than a chemical one.

There have been reports of molybdenum-reducing bacteria from agricultural soil, with a second molybdenum-reducing bacterium capable of degrading pollutant [4]. Molybdenum reduction to molybdenum blue is a mechanism related to bacterial growth and, like bacterial growth, exposes a peculiar stage in which the true growth rate begins at zero and quickly accelerates to a maximum value (max), resulting in a lag period (λ) [6]. Microbes have an incredible ability to resist the toxicity of heavy metals, which is useful in bioremediation. Ruminants are highly poisonous to it, and bioremediation has been reported in previous research [5]. Microbes have been responsible for molybdate decrease for over a century. In 1985, E. coli K12 was used in the first comprehensive study of molybdate reduction to Mo-blue. T. ferrooxidans, Enterobacter cloacae strain 48 (EC 48), and Serratia marcescens strain Dr.Y6 followed in 1988, 1993, and 2008 respectively [3,8].

MATERIALS AND METHODS

Data acquisition

The graphical data from Fig. 1. of the published work by Shukor et al. [6] of Mathematical Modeling of the Molybdenum Blue Production from Serratia sp. strain DRY5 was processed using the software Webplotdigitizer 2.5 [9], which helps to digitizes the scanned figure and has been widely used and acknowledged by many researchers because of its precision and reliability [3,5,6,10]. CurveExpert Professional software (Version 2.6.5) was used to model the data after been processed.

Fitting of the data

Employing CurveExpert Professional software (Version 2.6.5), nonlinear regression using the Marquardt algorithm was utilised to fit the bacterial growth curve using multiple growth models (Table 1). The algorithm looks for the most efficient way to lower the sum of squares between predicted and measured values. The sharpest gradient search of the curve between the four datum points was used to estimate the growth rate(max), while the line crossing the X-axis was used to estimate the lag time (λ). The highest growth rate was picked for the modelling exercise.

Table 1. Mo-blue production models employed in this research.

Model	Equation	No. Of	Ref.
		parameters	
Monod	S	2	[15]
	$\mu_{max} \overline{K_a + S}$		
Haldane	S	3	[16]
	μ_{max} (S ²)		[-•]
	$S + K_s + \left(\frac{S}{K_i}\right)$		
Teissier	1 1 1 1 1 1 1 1 1 1	3	[17]
	$\mu_{max} \left\{ 1 - exp \left(\frac{1}{K_i} \right) \right\}$		
	(S)		
	$-\exp\left(\frac{1}{K_{o}}\right)$		
Luong	S S	4	[18]
8	$\mu_{max} \frac{1}{K_{a} + S} (1 - \frac{1}{S_{m}})^{n}$		L - J
Aiba	$S \left(\frac{-S}{-S}\right)$	4	[19]
	$\mu_{max} \frac{1}{K_r + S} exp^{\kappa_i}$		L · J
Yano and	$\mu_{max}S$	4	[20]
Koga	(S^2) (S^2)		L . J
8	$S + K_s + \left(\frac{1}{K_i}\right) \left(1 + \frac{1}{K}\right)$		
Han and	$\left[\left(c \right) \right]^{n}$	5	[21]
Levenspiel	$q_{\max}S\left[1-\left(\frac{S}{S_m}\right)\right]$		
	$q = \frac{\left[\left(c \right) \right]^m}{\left[\left(c \right) \right]^m}$		
	$K_s + S - \left[1 - \left(\frac{S}{S_m}\right)\right]$		

Note:

 μ_{max} = maximum cell growth and degradation rate (h⁻¹) K_s = Half saturation constant or half velocity constant (% v/v)

 K_i = Inhibition constant (% v/v)

S = Substrate concentration (% v/v) S_m = Maximum concentration of substrate tolerated (% v/v)

m, n, K = curve parameters

Statistical analysis

The adjusted determination coefficient (R²), accuracy factor (AF), bias factor (BF), root-mean-square error (RMSE), and aicc (akaike information criterion) corrected were used to calculate the statistically significant difference between the models, as previously described [4,11–13]. The first to propose and suggest BF and AF were ross and McMeekin [14].

RESULTS AND DISCUSSION

Based on bacterial growth modelling (Figs 1-7), the Monod model was found to be the best model, with the lowest RMSE, AICc, and adjustedR² values. The model's AF and BF values were similarly excellent, with values close to 1.0. Except for the Tessier-Edward model, which shows poorest match (Table 2), the coefficients parameters for the Monod model were provided in Table 3.

Table 2. Statistical analysis of the various models used in this sturdy.

Model	Р	RMSE	adR2	AICc	BF	AF
Luong	4	0.1503	0.952	-37	0.680	1.598
Yano	4	0.1364	0.959	-40	1.312	1.440
Tessier-Edward	3	0.4252	-0.941	-11	1.089	1.353
Aiba	3	0.1551	0.947	-41	0.764	1.460
Haldane	3	0.1457	0.954	-43	0.707	1.546
Monod	2	0.1595	0.944	-45	0.692	1.625
Han and Levenspiel	5	0.1586	0.9454	-29	0.687	1.625
Note:						
P=number of parameters						

Table 3. Growth coefficients as predicted by the Monod model.

	μ_{m}	Ks		
	(h ⁻¹)	(% v/v)		
Value	3.86	43.41		
Std err	0.7269	14.371		
Range (95% confidence)	2.29 to 5.43	12.36 to 74.46		



Fig. 1. Growth of Serratia sp. strain DRY5 modeled using Tessier-Edward



Fig. 3. Growth of Serratia sp. strain DRY5 modeled using Hanlevenspiel



Fig. 5. Growth of Serratia sp. strain DRY5 modeled using Yano.



Fig. 2. Growth of Serratia sp. strain DRY5 modeled using Haldane



Fig. 4. Growth of Serratia sp. strain DRY5 modeled using Aiba.



Fig. 6. Growth of Serratia sp. strain DRY5 modeled using Monod.



Fig. 7. Growth of Serratia sp. strain DRY5 modeled using Luong.

CONCLUSION

In conclusion, based on statistical analyses such as corrected AICc (Akaike Information Criterion), bias factor (BF), adjusted coefficient of determination $(adjR^2)$, and root-mean-square error (RMSE), the Monod model was the best model in modelling the growth of *Serratia* sp in the production of Molybdenum Blue). Maximum specific growth rate (qmax) and Ks (concentration of substrate at half maximal specific growth rate (mg/L) were obtained from the fitting exercise, with values of 3.86 (95% confidence interval of 2.29 to 5.43) and 43.41 (95% confidence interval of 12.36 to 74.46), respectively. Under batch experiments, these biologically meaningful coefficients will be effective in predicting *Serratia sp.* strain DRY5 growth requirements in the production of Molybdenum Blue and therefore heavy metal will be remediated from the environment.

REFERENCES

- Idris D, Gafasa MA, Ibrahim SS, Babandi A, Shehu D, Ya'u M, et al. Pantoea sp. strain HMY-P4 Reduced Toxic Hexavalent Molybdenum to Insoluble Molybdenum Blue. J Biochem Microbiol Biotechnol. 2019;7(1):31–7.
- Othman AR, Bakar NA, Halmi MIE, Johari WLW, Ahmad SA, Jirangon H, et al. Kinetics of molybdenum reduction to molybdenum blue by *Bacillus* sp. strain A.rzi. Biomed Res Int. 2013;2013.
- Shukor MY, Rahman MF, Suhaili Z, Shamaan NA, Syed MA. Bacterial reduction of hexavalent molybdenum to molybdenum blue. World J Microbiol Biotechnol. 2009;25(7):1225–34.

- Halmi M, Wan Johari W, Mohd Ali M, Shaharuddin N. Isolation of molybdenum-reducing bacterium; *Serratia* sp. Strain MIE2 from agriculture soil and its potential use in soil bioremediation. J Biochem Microbiol Biotechnol. 2017;5(2):12–8.
- Ahmad SA, Shukor MY, Shamaan NA, Mac Cormack WP, Syed MA. Molybdate reduction to molybdenum blue by an antarctic bacterium. Biomed Res Int. 2013;2013.
- Syed MA, Shamaan NA, Shukor MY. Mathematical Modeling of the Molybdenum Blue Production from *Serratia sp*. strain DRY5. J Environ Microbiol, 2021;8(2):12-17.
- Yakasai HM, Rahman MF, Gusmanizar N, Shukor My. Mathematical Modeling of Molybdenum-Blue Production from Bacillus sp. strain Neni-10. Bioremed Sci Technol Res, 2021;9(1):7–12.
- Ariff AB, Rosfarizan M, Ghani B, Sugio T, Karim MIA. Molybdenum reductase in *Enterobacter cloacae*. World J Microbiol Biotechnol. 1997;13(6):643–7.
- Rohatgi A. WebPlotDigitizer User Manual 4.3. (http://arohatgi.info/WebPlotDigitizer/app/ Accessed June 2 2014). 2020;1–17.
- Halmi MI, Shukor MS, Johari WL, Shukor MY. Mathematical modeling of the growth kinetics of *Bacillus* sp. on tannery effluent containing chromate. J Environ Bioremed Toxicol. 2014;2(1):6-10.
- Motulsky HJ, Ransnas LA. Fitting curves to data using nonlinear regression: a practical and nonmathematical review. FASEB J. 1987;1(5):365-74.
- Yunus M, Shukor A, Sabo IA, Yahuza S, Dan-iya BI, Adamu S. Prediction of Cumulative Death Cases in The United States Due to COVID-19 Using Mathematical Models. J Environ Microbiol, 2020;8(1):37–41.
- Yahuza S, Sabo IA, Abubakar A, Yunus M, Shukor A. Thermodynamic Study on the Biosorptive Removal of Lead (II) Ions from Aqueous Solutions using Acid-treated *Cystoseira tricta* Biomass. J Environ Microbiol, 2020;8(2):1–5.
- 14. Ross T, McMeekin TA. Predictive microbiology. Int J Food Microbiol. 1994;23(3–4):241–64.
- Monod J.The growth of bacterial cultures. Annu Rev Microbiol. 1949;3(XI):371–94.
- 16. Haldane JBS. Enzymes, London, Longmans, Green. 1930.
- 17. Teissier G. Croissance des populations bacte riennes et quantite d'aliment disponible (Growth of bacterial populations and the available substrate concentration). Revis Sci. 1942;80:209.
- Luong JHT. Generalization of Monod kinetics for analysis of growth data with substrate inhibition. Biotechnol Bioeng. 1987;29(2):242-8.
- Aiba S, Shoda M, Nagalani M. Kinetics of product inhibition in alcohol fermentation. Biotechnol Bioeng. 1968;10(6):845–864.
- Yano T, Koga S. Dynamic behavior of the chemostat subject to substrate inhibition. Biotechnol Bioeng. 1969;11(2):139–153.
- Han K, Levenspiel O. Extended Monod kinetics for substrate, product, and cell inhibition. Biotechnol Bioeng. 1988;32(4):430-47.